



Probing molecular architectures and interactions with scanning tunneling microscopy on graphite and arachidic acid functionalized surfaces

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Abstract. This study investigates the application of scanning tunneling microscopy in exploring molecular structures and interfaces relevant to nanotechnology. Graphite was selected as a sample due to its ease of visualization of atomic arrangements and surface inertness. The surface of highly oriented pyrolytic graphite treated with arachidic acid ($C_{20}H_{32}O_2$) was examined to gain insights into the behavior of organic molecules at liquid-solid interfaces. Through detailed observations, the study demonstrates the versatility of scanning tunneling microscopy in elucidating molecular architectures and interactions. The findings underscore the importance of scanning tunneling microscopy in advancing our understanding of molecular systems and driving progress in nanotechnology applications. This work highlights the pivotal role of scanning tunneling microscopy in unraveling the complexities of nanoscale phenomena and fostering future innovations in the field.

Keywords: scanning tunneling microscopy, graphite, molecular structures, liquid-solid interfaces, nanotechnology applications

1. Introduction

Solid surface atomic composition can be observed with a Scanning Tunneling Microscope (*STM*). These compositions affect the dynamics of the solid, which makes them significant in material physics [1-2]. Quantum tunneling is the foundation of *STM* operation. The quantum mechanical tunneling effect is the fundamental idea that drives *STM* [3-4]. The ability of particles to pass over potential barriers that are taller than their total energy is described by this phenomena. The properties of particle waves are intimately related to the tunneling effect. A model of the energy levels of free electrons in a metal can explain this phenomena [5-6]. The energy of electrons can be expressed by the formula:

$$E = \frac{p^2}{2m} \quad (1)$$

Where, p is the electron's momentum and m is its mass, because the electron gas inside the conductor is regarded as free in this model. The Fermi level is the greatest energy an electron can have in a metal at zero temperature [7]. Conduction electrons could potentially find a trench throughout the metal's entire volume. The electrons with the highest energy, or those around Fermi level, are the ones that primarily contribute to the tunnel current. Conduction electrons are located at the edge of the potential well, which acts as a potential barrier for them at the metal surface and the boundary between the metal and vacuum. The output work ϕ determines the height of this potential barrier.

STM under ultrahigh vacuum conditions allows obtaining atomic images of the surface of metals and semiconductors, including silicon, by registering the tunneling current between the probe and the surface under study close to it. It is designed to study elementary structural processes on the surface of semiconductors and metals to create nanosystems and new materials with unique

properties. Not only may *STM* be used in ultra-high vacuum, but it can also be employed at temperatures ranging from almost 0 K to more than 1000 °C in air, water, and other liquids or gases with ambient temperatures [8].

Many of the present and future uses of nanotechnology, such as molecular light-harvesting and emitting devices, molecular electronics, biological identification, and molecular sensor technologies, are based on organic molecules [9]. Systems based on molecular thin films are emerging as a very promising path for the near future, whereas single-molecule electronics pose considerable obstacles for practical applications. One prominent technique for developing new molecular architectures is self-assembly, especially when it comes to the production of electrical and optoelectronic devices.

2. Methods

In this work, graphite was chosen as a sample for *STM* studies due to the ease of visualization of its atoms and the inertness of its surface. Since each crystallite in synthetic highly oriented pyrolytic graphite (*HOPG*) is perfectly aligned, it is utilized in place of natural graphite. The graphite topographic image presented in Figure 1 has a distinctive hexagonal pattern that is distinct from the honeycomb pattern [10]. This effect is caused by the electrical structure's quirks, which cause only every other atom to be mapped. Because the electron density is focused closer to the bulk, the carbon atoms on the surface, which are positioned above other atoms of the second carbon layer, are not visible [11].

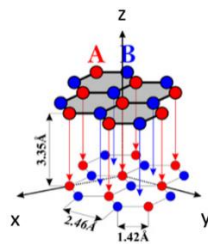


Figure 1 – Graphite's 3D atomic structure

When an electric field is applied to an extremely sharp metal tip in close proximity to an electrically conductive sample, a current flows between the tip and the sample without any physical contact. Using this so-called tunneling current, the electrical topography of the surface of freshly produced graphite is examined on the supplementary nanometer scale. Atoms and hexagonal structures are photographed by scanning the graphite surface line by line with the tip line. The main part of the equipment is scanning, based on a platform for vibration isolation and a protective magnetic cover for tip and sample inspection. Without the necessity for mechanical contact, a current can be produced between a sharp metal tip and an electrically conductive sample when an electric field is introduced. This phenomenon, called tunneling current, is utilized to examine the surface electron topography of freshly manufactured graphite at an additional nanoscale scale. Atoms and hexagonal structures are photographed through the application of a tip line over the graphite surface.

Figure 2 *a* presents a piezo-elements enable metal tip, which move across a sample's surface during *STM*.

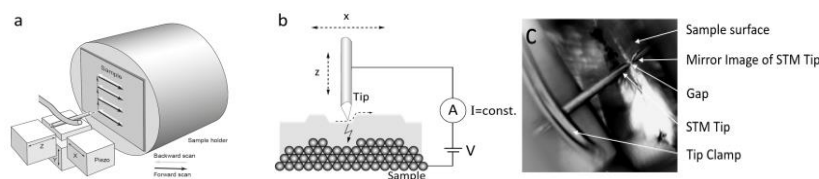


Figure 2 – *STM* sample mounted scheme

In Figure 2 *b* and *c* the tunneling current at which the tip is held is determined by the feedback loop *z*'s operational point. In essence, this establishes the separation between the tip and the model. The tip is nearer the sample surface the higher the set point. 5 nA is typically utilized for graphite atomic resolution.

By varying the voltage, these elements can precisely move the tip at a scale of picometers since they change length in reaction to applied voltage. A tunneling current is created across the tiny gap between the tip and the sample by an applied bias voltage. Constant-current or constant-height modes can be used for surface scanning. The tunneling current in the constant-height mode changes according to the tip's lateral position. The more widely used constant-current mode produces a profile of the sample's height based on the *z*-signal by using a feedback loop to maintain the tip height and keep the tunneling current constant. The tunneling current and distance have an exponential connection (a change of one Angström in distance results in a tenfold change in tunneling current) and the piezo-elements' exact movement allows for an extraordinary resolution, all the way down to the atomic level.

The *P*, *I* and *D* coefficients used in proportional-integral-differential (*PID*) controllers determine how the system reacts to deviations of the measured current from the set point. The number of values that are too low can cause the tip to be insufficiently sensitive to changes in height, while values that are too high can cause tip position fluctuations. Recommended parameter *D* can be set to zero. The values of parameters *P* and *I* was selected based on the current context and system response time. The default value of 1000 is a precise starting value.

The tip voltage indicates the offset applied to the tip. The lower the voltage, the closer the tip is to the surface. The optimum value of the bias voltage depends on the electronic structure of the sample, such as the density of states (*DOS*). For semiconductors, it is important to keep in mind that it is not recommended to place the bias voltage inside the band of forbidden energy states.

The *HOPG* surface was scanned at *STM* in air, which allowed imaging of different sizes, ranging from ~ 100 nm, where step edges are visible, to as small as ~ 5 nm with atomic resolution. The following characteristics of 5 nA, 0.6 nA, 1.3 V have been used for atomic resolution.

For more detailed analysis of a liquid-solid interface, we investigated the surface of *HOPG* treated with arachidic acid $C_{20}H_{32}O_2$.

3. Results and Discussion

Figure 3 presents the topography surface of the synthetic graphite scanned by *STM*. A scanning probe microscope provides an image of a surface magnified in all three dimensions: *x*, *y* and *z*, with maximum resolution in each axis. *x* –axis have been selected 1.2 nA, P_{gain} 1200, I_{gain} 1500., with the maximum resolution for each axis determined by various factors. The *z*-axis resolution is limited, first, by the sensitivity of the sensor and, second, by the amplitude of the probe vibrations relative to the sample surface. The design of the microscope must be able to reduce the amplitude of these vibrations to fractions of an angstrom.

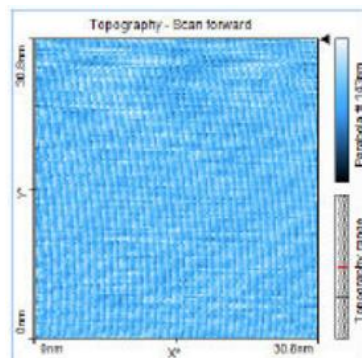


Figure 3 – Topography surface of 30 nm, scanned in 0.14 s, with 1.2 nA, P_{gain} 1200, I_{gain} 1500

Since thermal fluctuations have an impact on nanoscale measurements, we have selected a mode of 0.03 s at 128 "Points/Line" for atomic resolution in order to scan the sample as quickly as possible. Since that four to eight atoms have a diameter of one nanometer, much smaller picture sizes are required to reach atomic resolution. On Figure 4, atomic arrangements are often identifiable at image sizes of roughly 10–3 nm. Consequently: In the imaging panel, set the image size to 3 nm (Figure 5).

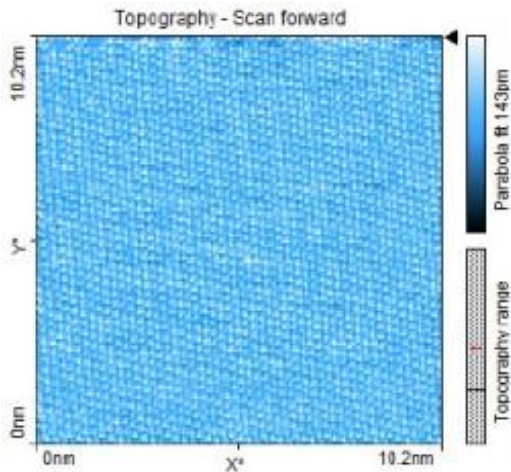


Figure 4 – Topography surface of 10 nm, scanned in 0.03 s, with 1.2 nA, P_{gain} 1200, I_{gain} 1500

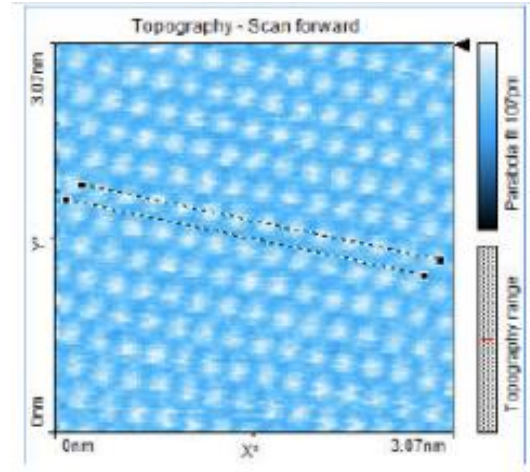


Figure 5 – Topography surface of 3 nm, scanned in 0.03 s, with 1.2 nA, P_{gain} 1300, I_{gain} 850. Line-to-line distance: $d=138$ pm

To investigate the interface between the liquid-solid interface, we applied arachidic acid $C_{20}H_{32}O_2$ to the *HOPG*. Figure 6a shows the condition of the test sample applied with arachidic acid. Parameters for scanning were as follows 0.6 nA, 1.3 V, image size $15.8 \times 15.8 \text{ nm}^2$. The scanned image on Figure 6b showed that at fine resolution, the obvious features of arachidic acid, which has an orange coloration, dominate. However, at the same parameters, it is almost impossible to distinguish individual carbon atoms.

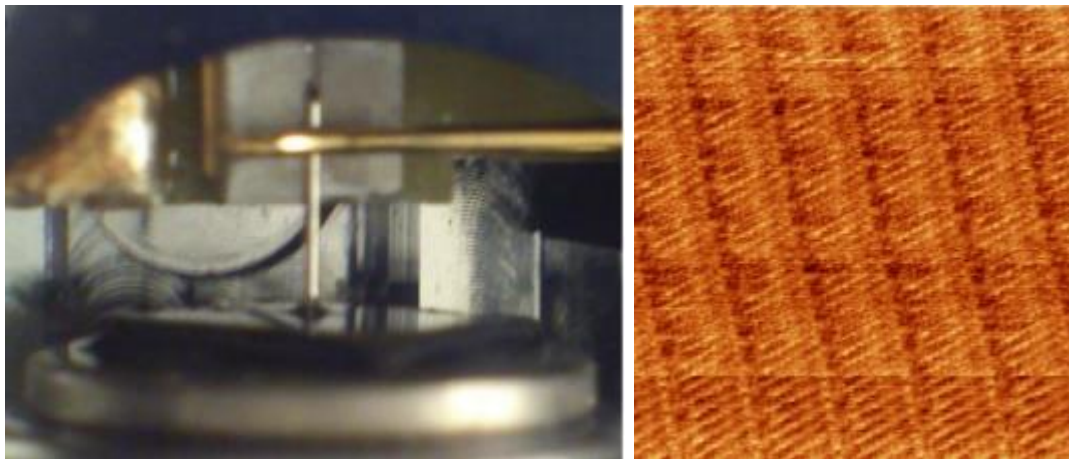


Figure 6 – Applying acid $C_{20}H_{32}O_2$ on *HOPG* surface: a) test sample; b) scanned image

4. Conclusions

In conclusion, this study demonstrates the versatility and potential of *STM* in investigating various interfaces and molecular structures. By utilizing graphite as a sample, the ease of visualizing atomic arrangements and the inert surface nature facilitated detailed observations. Furthermore, the exploration of the *HOPG* surface treated with arachidic acid provided insights into the behavior of

organic molecules at liquid-solid interfaces. These findings underscore the significance of *STM* in advancing our understanding of molecular structures and their interactions, paving the way for further developments in nanotechnology applications, such as molecular electronics, optoelectronic devices, and sensor technologies. Overall, this work highlights the crucial role of *STM* in unraveling the complexities of molecular systems and driving advancements in the field of nanoscience.

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